

SUPERIOR BEEF







SHEET DX0050 I DX0055

Salt Block - Salt Blocks - 98% sodium chloride

Elemental composition of a salt as measured by X-ray fl uoreszenzanalyse (RFA) Samples:

Number:

type: white natural salt

Designation: # 1_Salz

Amount: approximately 60 g
Sample receipt: on 2/21/02

Task: It should be specified with RFA an overview of the element or oxide components of the salt.

Measurement conditions: Method: X fl uoreszenzanalyse (RFA)

Device: Philips
Power: 2400 W
exciting wavelength: 0.615 Å, $Rh-k_\alpha$ Tension: 60kVanalyzer: LiF 220
Detector: Ge 111 PE PX1

Elements: z = 9 to 92 z = (fluorine to uranium) calibration: by CaO and model element

Detection limits: 100ppm bzw- 0.01wt%

Preparation: fine grinding

Evaluation: according to the elements and oxides gänigsten



Results: The following table following statements can be taken:

- The salt mainly consists of sodium chloride.
- In the 0.1% range are in the salt compounds of magnesium, aluminum, calcium, sulfur, vanadium and potassium.
- With the XRF method happen to be still detectable, the elements iron, Si, niobium, nickel and bromine
- All other elements are below the detection limit of 0.01 wt%. To better understand the element symbols a copy of the periodic table is attached.

Element / oxide concentration [% wt] # 1_Salz		
F	<00:01	
N/A	35	
MgO	12:19	
Al ₂ O ₃	12:07	
SiO ₂	12:03	
P ₂ O ₅	<00:01	
SO₃	12:31	
CI	63	
Ar	<00:01	
К	12:14	
CaO	12:34	
TiO ₂	<00:01	
V ₂ O ₅	12:21	
Cr ₂ O ₃	<00:01	
MnO	<00:01	
Fe ₂ O ₃	12:03	
C0 ₃ O ₄	<00:01	
NiO	12:02	
CuO	<00:01	
ZnO	<00:01	
Ga₂O₃	<00:01	
GeO₂	<00:01	
As ₂ O ₃	<00:01	
SeO ₂	<00:01	
br	12:02	
Rb₂O	<00:01	
SrO	<00:01	
Y ₂ O ₃	<00:01	
ZrO ₂	<00:01	
Nb ₂ O ₃	12:02	
MoO ₃	<00:01	

Rh₂O₃ <00:01 PdO <00:01 Ag₂O <00:01 CdO <00:01 In₂O₃ <00:01 SnO₂ <00:01 Sb₂O₃ <00:01 TeO₂ <00:01 J <00:01 BaO <00:01 La₂O₃ <00:01 Ce₂O₃ <00:01 Pr₂O₃ <00:01 Nd₂O₃ <00:01 Sm₂O₃ <00:01 Tb₂O₃ <00:01 Tb₂O₃ <00:01 HfO₂ <00:01 Tra₂O₃ <00:01 WO₃ <00:01 Pt <00:01 Au <00:01 HgO <00:01 Ti₂O₃ <00:01 PbO <00:01 Bi₂O₃ <00:01	Element / oxide concentration [% wt] # 1_Salz	
Ag ₂ O <00:01	Rh₂O₃	<00:01
CdO <00:01	PdO	<00:01
In₂O₃ <00:01	Ag ₂ O	<00:01
SnO₂ <00:01	CdO	<00:01
Sb ₂ O ₃ <00:01	In₂O₃	<00:01
TeO2 <00:01	SnO ₂	<00:01
J <00:01	Sb ₂ O ₃	<00:01
Cs2O <00:01	TeO₂	<00:01
BaO <00:01	J	<00:01
La ₂ O ₃ <00:01	Cs ₂ O	<00:01
Ce2O3 <00:01	BaO	<00:01
Pr ₂ O ₃ <00:01	La₂O₃	<00:01
$\begin{array}{c cccc} Nd_2O_3 & <00:01 \\ Sm_2O_3 & <00:01 \\ Gd_2O_3 & <00:01 \\ Tb_2O_3 & <00:01 \\ Tb_2O_3 & <00:01 \\ HfO_2 & <00:01 \\ Ta_2O_5 & <00:01 \\ WO_3 & <00:01 \\ Ir & <00:01 \\ Pt & <00:01 \\ Au & <00:01 \\ HgO & <00:01 \\ Ti_2O_3 & <00:01 \\ \hline PbO & <00:01 \\ \hline Bi_2O_3 & <00:01 \\ \hline \end{array}$	Ce ₂ O ₃	<00:01
Sm ₂ O ₃ <00:01	Pr₂O₃	<00:01
$\begin{array}{c cccc} Gd_2O_3 & <00:01 \\ Tb_2O_3 & <00:01 \\ HfO_2 & <00:01 \\ Ta_2O_5 & <00:01 \\ WO_3 & <00:01 \\ Ir & <00:01 \\ Pt & <00:01 \\ Au & <00:01 \\ HgO & <00:01 \\ Ti_2O_3 & <00:01 \\ PbO & <00:01 \\ \end{array}$	Nd ₂ O ₃	<00:01
Tb ₂ O ₃ <00:01 HfO ₂ <00:01 Ta ₂ O ₅ <00:01 WO ₃ <00:01 Ir <00:01 Pt <00:01 Au <00:01 HgO <00:01 Ti ₂ O ₃ <00:01 PbO <00:01 Bi ₂ O ₃ <00:01	Sm ₂ O ₃	<00:01
HfO2 <00:01	Gd₂O₃	<00:01
Ta ₂ O ₅ <00:01	Tb ₂ O ₃	<00:01
WO ₃ <00:01 Ir <00:01 Pt <00:01 Au <00:01 HgO <00:01 Ti ₂ O ₃ <00:01 PbO <00:01 Bi ₂ O ₃ <00:01	HfO ₂	<00:01
Ir < <00:01 Pt < <00:01 Au < <00:01 HgO < <00:01 Ti ₂ O ₃ < <00:01 PbO < <00:01 Bi ₂ O ₃ < <00:01	Ta₂O₅	<00:01
Pt < 00:01 Au < 00:01 HgO < 00:01 Ti ₂ O ₃ < 00:01 PbO < 00:01 Bi ₂ O ₃ < 00:01	WO₃	<00:01
Au <00:01 HgO <00:01 Ti ₂ O ₃ <00:01 PbO <00:01 Bi ₂ O ₃ <00:01	Ir	<00:01
HgO <00:01 Ti ₂ O ₃ <00:01 PbO <00:01 Bi ₂ O ₃ <00:01	Pt	<00:01
Ti ₂ O ₃ <00:01 PbO <00:01 Bi ₂ O ₃ <00:01	Au	<00:01
PbO <00:01 Bi ₂ O ₃ <00:01	HgO	<00:01
Bi ₂ O ₃ <00:01	Ti₂O₃	<00:01
	PbO	<00:01
	Bi ₂ O ₃	<00:01
ThO ₂ <00:01	ThO₂	<00:01
U ₃ O ₈ <00:01	U₃O ₈	<00:01